

MUHAMMAD USAMA JAMAL

Energy transfer to luminescence  
centres in alkali and  
rare-earth metal molybdates





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This study was carried out at the Institute of Physics, University of Tartu, Estonia.

The dissertation was admitted on June 02, 2026, in partial fulfilment of the requirements for the degree of Doctor of Philosophy in Physics, and was allowed for defence by the Scientific Council of the Institute of Physics, University of Tartu.

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Defence: August 25, 2026, at University of Tartu, Estonia

This work has been supported by Estonian Research Council grant PRG2733, the ERDF funding in Estonia granted to the Centre of Excellence TK210 “Centre of Excellence in Sustainable Green Hydrogen and Energy Technologies” and TWINNING project “EXANST” funded by the European Union under grant agreement No. 101159716.



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ISSN 1406-0647 (print)  
ISBN 978-9908-57-257-4 (print)  
ISSN 2806-2523 (pdf)  
ISBN 978-9908-57-258-1 (pdf)

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University of Tartu Press  
[www.tyk.ee](http://www.tyk.ee)

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## LIST OF PUBLICATIONS INCLUDED IN THE THESIS

- I. **M.U. Jamal**, V. Nagirnyi, H. Mändar, K. Chernenko, D. Spassky, Structural and luminescence properties of  $K_5Nd(MoO_4)_4$  and  $KNd_5(MoO_4)_8$ , *Journal of Luminescence* (2026) 121995. <https://doi.org/10.1016/j.jlumin.2026.121995>.
- II. **M.U. Jamal**, V. Nagirnyi, K. Chernenko, A. Kotlov, Y. Smortsova, D. Spassky, Crystal structure controlled energy transfer to  $Tb^{3+}$  in  $KTb(MoO_4)_2$  and  $K_5Tb(MoO_4)_4$  crystals, *Materials Research Bulletin* 191 (2025) 113553. <https://doi.org/10.1016/j.materresbull.2025.113553>.
- III. D. Spassky, A. Vasil'ev, **M.U. Jamal**, V.A. Morozov, B.I. Lazoryak, B.S. Redkin, K. Chernenko, V. Nagirnyi, Temperature dependent energy transfer to  $Eu^{3+}$  emission centres in  $K_5Eu(MoO_4)_4$  crystals, *CrystEngComm* 26 (2024) 1106–1116. <https://doi.org/10.1039/D3CE01201H>.

## AUTHOR'S CONTRIBUTION

- Publication I. The author contributed to setting the physical problem and determining the strategy for conducting experimental research, writing the original draft, manuscript review and editing, experimental work, figure preparation, and data processing and analysis.
- Publication II. The author contributed to writing the original draft, manuscript review and editing, experimental work, figure preparation, and data processing and analysis.
- Publication III. The author contributed to manuscript review and editing, experimental work, figure preparation, and data processing and analysis.

## ABBREVIATIONS

e-h	electron-hole
K5Eu	$\text{K}_5\text{Eu}(\text{MoO}_4)_4$
K5Nd	$\text{K}_5\text{Nd}(\text{MoO}_4)_4$
K5Tb	$\text{K}_5\text{Tb}(\text{MoO}_4)_4$
KNd5	$\text{KNd}_5(\text{MoO}_4)_8$
KTb	$\text{KTb}(\text{MoO}_4)_2$
MEE	Multiplication of Electronic Excitation
NIR	Near-Infrared
OPO	Optical Parametric Oscillator
PL	Photoluminescence
PLE	Photoluminescence Excitation
PMT	Photomultiplier Tube
RE	Rare-Earth
RT	Room Temperature
STE	Self-Trapped Exciton
UV-VIS	Ultraviolet-Visible
VUV	Vacuum Ultraviolet

# 1. INTRODUCTION

Double molybdates containing alkali-metal and rare-earth ( $\text{RE}^{3+}$ ) ions have attracted considerable attention due to their promising luminescence, optical, lasing, magnetic, and scintillation properties [1] [2] [3] [4] [5] [6] [7] [8] [9] [10]. These materials constitute a broad and versatile class of functional inorganic compounds. Their structural flexibility, combined with the unique 4f electronic configurations of  $\text{RE}^{3+}$  ions, makes them highly suitable for a wide range of photonic and optoelectronic applications.

One of the key advantages of double molybdates is the high tunability of their luminescence properties [11] [12]. Emission and excitation behaviour can be effectively tailored through conventional compositional modification, such as substituting one rare-earth ion for another with a different 4f electronic configuration. For example, replacing  $\text{Eu}^{3+}$  with  $\text{Tb}^{3+}$  results in a shift of emission colour from red to green, arising from their distinct electronic structures [11]. In addition, the variation of the alkali-metal cation can significantly influence the crystal structure and the local coordination environment of  $\text{RE}^{3+}$  ions, thereby modifying the optical response [13].

Beyond the elemental substitution, luminescence properties can also be tailored by modifying stoichiometric ratio of the same elements. Such variations give rise to different crystal structures, leading to changes in the coordination of RE ions, site symmetry, structural ordering, and RE-RE distances. These structural differences play a crucial role in determining energy-transfer pathways and emission efficiency. Compared to the conventional substitution strategies, this approach remains relatively underexplored.

In this work, both strategies for tuning luminescence in double molybdates are systematically investigated. The first approach involves rare-earth ion substitution within the  $\text{K}_5\text{RE}(\text{MoO}_4)_4$  system, where  $\text{Eu}^{3+}$  is replaced by  $\text{Tb}^{3+}$  or  $\text{Nd}^{3+}$ . For the second approach, stoichiometrically distinct but elementally identical crystals are compared, specifically  $\text{K}_5\text{Tb}(\text{MoO}_4)_5$  ( $\text{K}_5\text{Tb}$ ) with  $\text{KTb}(\text{MoO}_4)_2$  ( $\text{KTb}$ ) and  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$  ( $\text{K}_5\text{Nd}$ ) with  $\text{KNd}_5(\text{MoO}_4)_8$  ( $\text{KNd}_5$ ). Although these compounds contain the same elements, they crystallize in different structures, leading to distinct luminescence behaviours.

A detailed understanding of how stoichiometric ratio driven structural variations influence luminescence properties, particularly in double molybdates, remains limited compared to that of elemental substitution, which is widely reported to tune the emission properties of these materials [11] [12] [14]. Therefore, in this work, a systematic investigation using both approaches is conducted to establish a structure-property relationship governing energy-transfer processes and luminescence properties of double molybdates, as well as the changes induced by  $\text{RE}^{3+}$  ion substitution. A comprehensive understanding of these relationships enables the development of more efficient, application-specific photonic materials.

To elucidate the changes in luminescence properties arising from these modifications, a systematic photoluminescence (PL) spectroscopy study is performed.

PL spectroscopy provides key information on emission wavelengths, excitation mechanisms, luminescence lifetimes, and thermal quenching behaviour, as well as insights into energy-transfer processes. The use of synchrotron radiation for luminescence spectroscopy enabled us to expand research into the deep VUV region.

In RE<sup>3+</sup> based double molybdates, excitation can occur via both intracentre and interband mechanisms. In intracentre excitation, electrons are excited within the 4f levels of the RE<sup>3+</sup> ions, followed by radiative or non-radiative relaxation to the lower emitting states. In interband excitation, energy is first absorbed by the MoO<sub>4</sub> complex, typically via transition of electrons from the 2p oxygen states forming the top of the valence band to the 4d molybdenum states forming the bottom of the conduction band, and subsequently transferred to the RE<sup>3+</sup> ions. During this process, intermediate phenomena such as inelastic scattering with creation of secondary e-h pairs, interactions with lattice vibrations (thermalization), exciton formation and charge-carrier trapping may occur. These processes noticeably influence the efficiency and pathway of energy transfer to the emitting centres. Therefore, understanding the energy transfer processes is essential for identifying the factors governing luminescence and for optimizing the optical performance of these materials.

The main aim of the thesis is:

- To investigate the influence of elemental composition and structural arrangement of potassium rare earth double molybdates on their luminescence properties, electronic band structure, and processes of energy transfer to the luminescence centres.

To achieve the aim, the following tasks were performed:

1. Measurement of XRD patterns to characterize the crystal.
2. Measurements of the absorption and reflection spectra to elucidate modifications of band structure and identify intracentre 4f–4f RE<sup>3+</sup> transitions.
3. Measurement of emission spectra over a wide UV-NIR spectral range to identify the characteristics of emission centres.
4. Photoluminescence excitation (PLE) spectroscopy in the 2.5–45 eV range to clarify the mechanisms of energy transfer to emission centres.
5. Determination of lifetimes of emitting states and investigation of thermal quenching processes using temperature-dependent decay kinetics and luminescence intensity measurements.

## 2. GENERAL BACKGROUND

### 2.1. Overview of energy transfer processes in luminescent materials

Energy transfer in luminescent ionic crystals is a multi-stage process initiated by interaction with ultraviolet (UV), vacuum ultraviolet (VUV), or ionizing radiation [15]. A pictorial representation of relaxation processes is shown in Fig. 1. Absorption of high-energy radiation generates primary, highly energetic charge carriers such as hot electrons excited high into the conduction band and holes created in the valence band or in core electronic states. Hot electrons undergo rapid inelastic electron–electron scattering, during which their primary excess kinetic energy is transferred to the electrons in the valence band, generating secondary electron-hole pairs in the material. The multiplication of electronic excitations continues until the electron energy drops below the electron-electron scattering threshold. Simultaneously, deep core holes relax through Auger processes. In this mechanism, an electron from the valence band moves to the core state hole, and the released energy is transferred non-radiatively to another electron, which is subsequently excited into the conduction band as an Auger electron. The combined inelastic scattering and Auger relaxation processes lead to rapid redistribution and multiplication of electronic excitations on ultrafast timescales of approximately  $10^{-15}$ – $10^{-16}$  s, while the total energy of the electronic subsystem remains nearly conserved.

The next stage is the thermalization of excited carriers through interaction with the crystal lattice. During this process, hot electrons and holes lose excess kinetic energy via phonon emission and relax toward the bottom of the conduction band and the top of the valence band, respectively. A substantial fraction of the absorbed energy is transferred to the lattice as vibrational energy.

After thermalization, depending on the material's lattice properties, the thermalized carriers may form self-trapped excitons (STEs), localize at defects, or be captured by impurity and luminescence centres during migration. These migration-stage interactions strongly influence the efficiency of energy transfer and luminescence formation. In rare-earth-doped materials, energy transfer from the host lattice to the 4f states of activator RE ions occurs via sequential capture of separated electrons and holes, or via the intermediate formation of excitonic states.

The final stage of energy relaxation involves radiative and non-radiative recombination. In radiative relaxation, the stored excitation energy is emitted as luminescence photons, whereas non-radiative processes dissipate energy into the lattice through phonon emission.

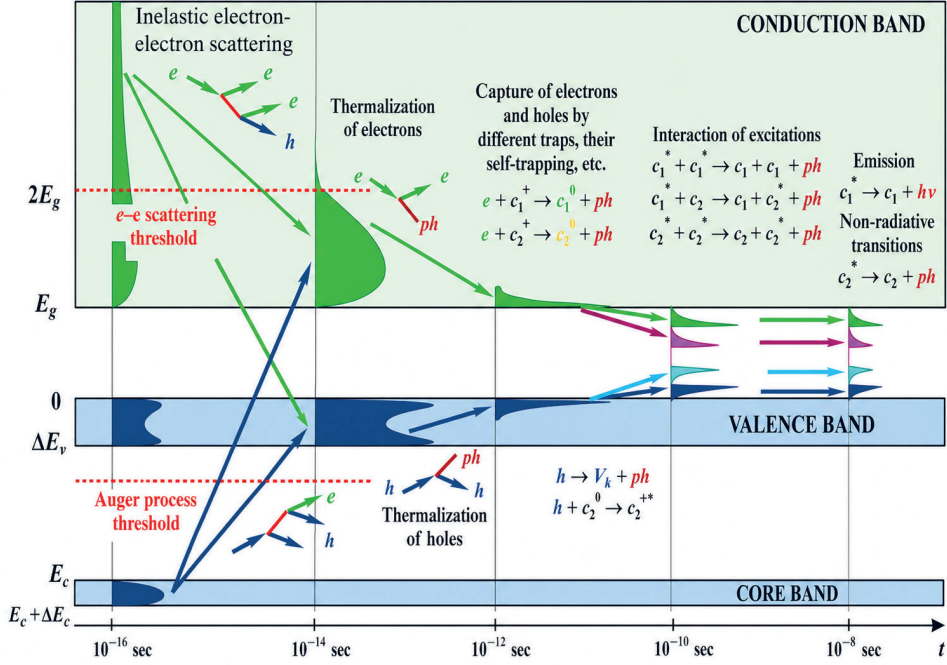


Figure 1: Scheme of relaxation of electronic excitations in an intrinsic scintillator.  $e$  = electrons,  $h$ =holes,  $ph$ =phonons,  $h\nu$ = photons,  $V_k$  = self-trapped holes,  $c_n$  = ionic centres with charge  $n$  [15].

## 2.2. Overview of double molybdates

To date, many double molybdate compounds have been synthesized and structurally characterized in detail [1] [11] [14] [16] [17] [18]. In addition, a scheme has also been derived for  $KRE(MoO_4)_2$  type double molybdates, providing a systematic description of the relative positions of the  $RE^{3+}$  energy levels with respect to the valence and conduction bands [19]. This scheme provides important insight into luminescence processes in double molybdates. In particular, the energy gap between the excited 5d levels of  $RE^{3+}$  ions and the bottom of the conduction band determines the thermal stability of the emission centres, since small gaps facilitate thermal ionization and luminescence quenching. Furthermore, analysis of the energy positions of the ground and excited 4f states helps clarify the mechanisms of host-to-RE energy transfer. When these 4f states are located close to or overlap with the valence or conduction bands, direct sequential capture of separated electrons and holes by the  $RE^{3+}$  ions becomes inefficient due to competing carrier thermalization processes in the lattice. Under these conditions, energy transfer from the host lattice to the rare-earth ions is expected to occur predominantly via intermediate excitonic states. According to the scheme proposed in [19], this mechanism is expected to dominate in the studied potassium double molybdates with  $RE = Eu, Tb, \text{ and } Nd$ .

The crystals studied in this work, such as  $\text{K}_5\text{Eu}(\text{MoO}_4)_4$  (K5Eu),  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$ ,  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$  and  $\text{KTb}(\text{MoO}_4)_2$ , have also been investigated previously from a crystallographic perspective, and information on phase formation, crystal symmetry, lattice parameters, and the coordination environments of the rare-earth ions is already known [8] [9] [11] [1] [20]. The luminescence properties of these compounds have not been investigated in detail. In many cases, general spectroscopic experiments were performed, including recording emission spectra under the most intense excitation line to illustrate the main emission from the samples, measuring its excitation spectra in the conventional UV-Vis range and luminescence decay kinetics. However, comprehensive studies of excitation mechanisms, energy transfer processes, site-selective luminescence, luminescence thermal quenching, electron relaxation mechanisms within the 4f states of  $\text{RE}^{3+}$ , and secondary excitation creation under high-energy excitation remain largely absent.

Among the different structural families of double molybdates, compounds crystallizing in the  $\text{MRE}_5(\text{MoO}_4)_8$  form remain particularly underexplored. Many possible alkali metal-rare-earth combinations in this structural type have not yet been reported, such as  $\text{KNd}_5(\text{MoO}_4)_8$ , leaving the influence of stoichiometric ratio driven structural variations on optical properties insufficiently understood. Therefore, significant gaps also remain in understanding how crystal structure, local symmetry, and RE-RE distances influence the optical and luminescence properties of these materials.

The present work aims to partially fill these gaps by conducting a comprehensive investigation of the luminescence properties and energy-transfer processes in selected potassium rare-earth molybdate crystals. Particular emphasis is placed on comparing intracentre and interband excitation processes and on analyzing how these excitation pathways affect emission spectra, luminescence decay kinetics, and the thermal stability of luminescence. Furthermore, to obtain deeper insight into the local environment of the rare-earth ions, high-resolution emission and excitation spectroscopy were employed. In addition, site-selective luminescence studies were performed to distinguish the optical behaviour of different rare-earth centres present in the crystals. Another important aspect of this work is the use of powerful excitation sources, including an optical parametric oscillator (OPO) laser and synchrotron radiation, for recording emission and excitation spectra over a broad energy range. Unlike conventional excitation sources such as xenon or deuterium lamps, these high-intensity excitation methods enable the observation of emissions from higher-lying 4f excited states that are normally inaccessible or too weak to detect. As a result, additional radiative and non-radiative relaxation channels can be identified and analysed in detail. Furthermore, excitation spectra measured under synchrotron radiation over the wide energy range of 4–45 eV provide important insight into high-energy excitation processes and energy-transfer mechanisms in these materials.

Overall, this work extended the current understanding of alkali rare-earth molybdates by moving beyond structural characterization to a detailed spectroscopic investigation of their luminescence properties and energy-transfer mechanisms.

## 3. METHODOLOGY

### 3.1. Preparation of the samples

The crystals investigated in this work were grown using the Czochralski technique under similar experimental conditions. Therefore, the detailed crystal growth procedure is described only for the KTb crystal as a representative example to illustrate the synthesis methodology employed for all the investigated crystals. The exact crystal growth details of the individual crystal can be seen in the following references [9] [11] [1]. The polycrystalline materials used to grow the KTb single crystal were synthesized by solid-state reaction. A stoichiometric mixture of the reagents ( $\text{K}_2\text{CO}_3$ ,  $\text{Tb}_4\text{O}_7$  of spectroscopic purity  $\geq 99.995\%$ , and  $\text{MoO}_3$ ) was annealed in a Pt crucible at 773–823 K for 15 h to decompose the carbonate. After that, the mixture was ground, homogenized, and heated at 1173 K for 15 h in air. The synthesized polycrystalline material was melted in a 40 mm diameter Pt crucible using an induction furnace operating at 17 kHz. A small-sized bar of non-oriented KTb single-crystal was used as a seed. The growing temperature was accurately determined by repeated seeding trials. A KTb colourless, transparent crystal with a diameter of up to 10 mm and a length of 20 mm was grown at 1373 K by the Czochralski technique at a pulling rate of 0.5–1.5 mm/h and a rotation rate of 50 rpm in a slightly oxidizing atmosphere [9].

### 3.2. Photoluminescence spectroscopy techniques

Photoluminescence spectroscopy is an essential experimental technique for investigating energy transfer processes and the luminescence properties of phosphors. It provides direct information on electronic transitions, emission mechanisms, and relaxation processes in luminescent materials. In this work, different PL spectroscopy setups were employed depending on the specific aim of each investigation. The details of these experimental setups are given below:

- The spectral measurements under excitation in the 2–6 eV energy range were performed at the UV-Vis setup in Tartu. Two types of lamps served as excitation sources: a 400-W deuterium discharge lamp (DDS-400) was used to measure excitation and emission spectra, whereas a PerkinElmer 150-W xenon pulsed flash lamp (1  $\mu\text{s}$ ) was used for decay kinetics. Both excitation sources are used in combination with a primary double-quartz prism monochromator (DMR-4). The samples were mounted in a Janis VPF-800 liquid nitrogen cryostat (65–800 K) either by conductive silver glue or by a screw-tightened copper clip. Emission spectra were recorded using a secondary ARC Spectra Pro 308i Czerny-Turner-type grating monochromator equipped with a photomultiplier tube H8259-01. Different SCHOTT glass filters were used to eliminate second-order excitation and emission in the secondary monochromator. Excitation and emission spectra were measured at room

temperature and at 78 K. A LakeShore 335 temperature controller was used to monitor temperature and control crystal heating. The experimental setup was operated using a LabView software.

- For excitation in the 3–11 eV region, a VUV system comprising a deuterium discharge lamp and a Seya Namioka McPherson 234/302 monochromator equipped with a 300 gr/mm grating (dispersion 16 nm/mm, optimized for 160 nm) was used. The samples were mounted on a sample holder of a closed-cycle helium cryostat from ARS (Advanced Research Systems, Inc.). The primary monochromator and cryostat were under a high vacuum of  $7 \times 10^{-7}$  mbar, maintained by a turbomolecular pump. The emission spectra were recorded using an Andor Shamrock 303i-B grating monochromator equipped with a CCD detector (Andor iDus 416) or with various photomultiplier tubes Hamamatsu H8259, H8259-01, or H8259-02. All measurements were carried out over the temperature range 5–350 K controlled using a LabView software.
- PL spectra and decay curves were also measured using a tunable optical parametric oscillator NT242, which provided excitation wavelength selection across 210–2600 nm with a 3-ns pulse duration. The samples were mounted in a Janis VPF-800 cryostat, allowing temperature control from 77 to 800 K using a Lake Shore 335 temperature controller. Depending on the measurement conditions, the emission spectra were recorded using either an Andor Shamrock 303i-B or a Kymera 328i spectrograph, coupled to a CCD detector. Decay curves were recorded using Hamamatsu R3377 and Hamamatsu H10330A-75 NIR-PMT detectors, sensitive in the UV-Vis and NIR regions, respectively, coupled to a LeCroy 6100A 1 GHz oscilloscope.
- Luminescence and PLE spectra with high spectral resolution were recorded using a specialized laboratory setup with an ARC 150 W Xe lamp as the excitation source. The excitation wavelength was selected using an MDR-206 monochromator equipped with a 1200 gr/mm grating (dispersion 4.2 nm/mm). The excitation spectra were measured with a spectral resolution of 0.4 nm. The samples mounted in a vacuum optical cryostat (Cryotrade LN-120) were studied over the temperature range of 77–500 K. Temperature was controlled using a LakeShore 335 controller. The luminescence spectra were recorded with a spectral resolution of 0.32 nm using an Oriel MS257 spectrograph with a 2400 gr/mm grating (dispersion 1.6 nm/mm), equipped with a Marconi 30–11 CCD detector.
- PLE spectra in the UV-VUV range (3.7–40 eV) were recorded at the P66 beamline of the PETRA III storage ring at DESY [21]. The excitation energy was selected using a 2 m McPherson monochromator equipped with two interchangeable holographic concave gratings (1200 gr/mm) coated with Pt and Al+MgF<sub>2</sub>. The Al+MgF<sub>2</sub> coated grating was optimized for the 3.75–11 eV spectral range, while the Pt-coated grating was used for measurements at higher excitation energies. The excitation spectra were measured with a

spectral resolution of approximately 0.5 nm (corresponding to  $\sim 0.06$  eV at 12 eV). The samples were mounted in a flow helium cryostat, which allowed them to cool to 10 K. The emitted luminescence was analysed using an ANDOR Kymera 328i monochromator and detected with a Hamamatsu R6358 photomultiplier tube. The PLE spectra were normalized to the luminescence yield of sodium salicylate.

- PL studies including emission, excitation and reflection spectra in the excitation energy range 4.5 – 45 eV were also conducted at the photoluminescence end station of the FinEstBeAMS beamline at the 1.5 GeV storage ring of the MAX IV synchrotron facility [22] [23]. The photon source for FinEstBeAMS is an in-house-built, elliptically polarized undulator (EPU) of the APPLE II type, which provides tunable photon energies from the ultraviolet to the soft X-ray range (4.5–1300 eV). Depending on photon energy, the photon flux varies from  $1 \times 10^{11}$  to  $8 \times 10^{13}$  ph/s. The generated beam is collimated by a toroidal mirror (M1) and directed to the input of a grazing-incidence primary monochromator equipped with a 92 lines/mm grating (See Fig.2), providing a photon-energy resolution of approximately 2.3 meV at 10 eV. To suppress higher-order excitation radiation from the grating, an appropriate set of optical filters was employed, depending on the first-order excitation energy. These included fused silica and  $\text{MgF}_2$  optical filters, as well as thin metallic films of In, Sn, Mg, and Al. The excitation spectra were subsequently corrected for the reference signal of a factory-calibrated AXUV-100G diode. The crystals were mounted in an ARS closed-cycle helium cryostat. To measure reflectivity spectra, the radiation reflected by the sample at an angle of  $22.5^\circ$  fell on a screen made of sodium salicylate, which has a constant light yield, and the salicylate emission was registered by a Hamamatsu H11123 counting head. PL excitation spectra were obtained by recording emission spectra with a CCD camera at different excitation energies, thereby covering the 300 nm range in a single scan. Time-resolved measurements were performed in single-bunch operation mode. In this mode, the duration of a single excitation pulse was  $\sim 160$  ps, with a 320 ns gap between pulses. The luminescence signal was recorded using an optical fiber-coupled Andor Shamrock SR-303i spectrometer equipped with a thermoelectrically cooled hybrid photodetector, HPM-100-07C, from Becker & Hickl. For steady-state measurements, a Newton CCD camera (Andor, model DU970P-BVF) was used in multi-bunch mode. The spectra were corrected for the registration system's spectral sensitivity function.

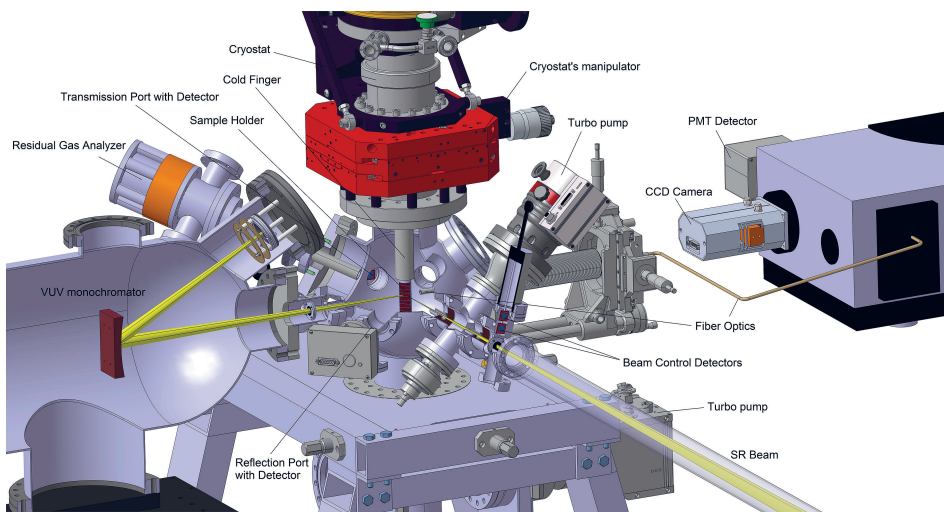


Figure 2: A scheme of the photoluminescence setup at the FinEstBeAMS beamline [22].

## 4. RESULTS AND DISCUSSION

This section presents a comparative study of the energy transfer mechanism and luminescence properties of  $\text{RE}^{3+}$  ions ( $\text{RE} = \text{Eu}, \text{Tb}, \text{Nd}$ ) in the  $\text{K}_5\text{RE}(\text{MoO}_4)_4$  system, along with crystals of identical elemental composition but different crystal structures, namely  $\text{KRE}(\text{MoO}_4)_2$  ( $\text{RE} = \text{Tb}$ ) and  $\text{KRE}_5(\text{MoO}_4)_8$  ( $\text{RE} = \text{Nd}$ ). The objective is to show how substituting  $\text{RE}^{3+}$  ions and varying the crystal structure influence the energy-transfer processes and luminescence properties of these ions.

A particular emphasis is placed on several key aspects. The fine structure of the emission lines is analysed, with special attention to how differences in the point symmetry of  $\text{RE}^{3+}$  and its nearest environment influence spectral line shapes and splitting patterns of emissions originating from the 4f states of  $\text{RE}^{3+}$  ions. Moreover, the influence of crystal structure on electronic band structure is investigated, providing insight into band structure engineering of the material. In addition, the effect of structural differences on luminescence decay kinetics is investigated, highlighting variations in radiative and nonradiative relaxation processes. The mechanisms responsible for thermal quenching of luminescence are also discussed, with a focus on how distinct lattice environments affect the thermal stability of the excited states.

### 4.1. Crystal structure

Most  $\text{M}_5\text{RE}(\text{MoO}_4)_4$  compounds crystallize in the R3m space group, similar to the  $\text{K}_2\text{Pb}(\text{SO}_4)_2$  type palmierite structure [24] [25]. Palmierite-related phases can be generally described by the formula  $\text{M}_2\text{M}_1(\text{AO}_4)_2$ , consisting of alternating M2 layers and  $[\text{M}_1\text{A}_2\text{O}_8]$  layers stacked along the c-axis of the rhombohedral unit cell. In this structure, the M2 and M1 cations are arranged in distinct layers. The M2 sites are exclusively occupied by  $\text{M}^+$  cations, whereas the M1 sites are occupied by either  $\text{RE}^{3+}$  ions or  $\text{M}^+$  ions in statistical alternation depending on the elemental composition. The crystal structure may exhibit various distortions depending on the constituent element. These distortions are mainly associated with the rotations of the  $\text{MoO}_4^{2-}$  tetrahedra and with the ordering of  $\text{RE}^{3+}$  and  $\text{M}^+$  cations at the M1 sites. Due to the structural flexibility of the  $\text{MoO}_4$  tetrahedral units, different coordination environments for  $\text{RE}^{3+}$  ions can be stabilized within the lattice. Ordering of  $\text{RE}^{3+}$  and  $\text{M}^+$  cations at the M1 sites plays a crucial role in determining the overall crystal symmetry. In particular, ordering between  $\text{M}^+$  and  $\text{RE}^{3+}$  cations can reduce symmetry from the ideal rhombohedral R3m to lower symmetry. Depending on the degree and nature of cation ordering, the structure may transform into a monoclinic phase (e.g., C2/c) or even into an incommensurately modulated monoclinic phase described by a superspace group such as  $\text{X}2/\text{m}(0\beta 0)00$ . Such symmetry-lowering due to cation ordering at the M1 site demonstrates that the palmierite-type structure is highly flexible, allowing it to accommodate cations of different sizes while maintaining structural stability.

It has been shown that the crystal structure of both K5Eu and K5Tb crystals can be refined using the Rietveld method within the framework of a (3+1)D incommensurately modulated structure described by the superspace group  $C2/m(0\beta 0)$  [11] [1]. In K5Eu, the local environments of K1 and  $\text{Eu}^{3+}$  sites are only weakly affected by the rotations of the  $\text{MoO}_4^{2-}$  tetrahedra and shifts of oxygen atoms, resulting in eightfold coordination for both K1 and  $\text{Eu}^{3+}$ . In contrast, the local structure around the K1 and  $\text{Tb}^{3+}$  sites in K5Tb is strongly influenced by the rotations of the  $\text{MoO}_4^{2-}$  tetrahedra and displacements of oxygen atoms. Consequently, the coordination environments are significantly distorted, leading to coordination numbers ranging from 6 to 10. In  $\text{K}_5\text{RE}(\text{MoO}_4)_4$ , the distance between the two nearest  $\text{RE}^{3+}$  is approximately 6 Å [8] [1].

$\text{KTb}(\text{MoO}_4)_2$  has an orthorhombic structure, indexed in the Pbcn space group [9] [14]. The structure consists of layers, formed by the  $[\text{TbMo}_2\text{O}_8]$  units, which include the  $\text{TbO}_8$  polyhedra and  $\text{MoO}_4$  tetrahedra. These layers are oriented perpendicular to the b-axis and are separated by zigzag layers of  $\text{K}^+$  ions. Neighboring  $\text{TbO}_8$  polyhedra share edges, forming infinite  $[\text{Tb}_2\text{O}_{14}]_n$  chains along the c-axis, with a Tb-Tb distance of about 4 Å within the chains [9] [16]. The  $\text{MoO}_4$  tetrahedra separate these chains, causing a one-dimensional arrangement of  $\text{Tb}^{3+}$  ions.

Powder X-ray diffraction characterization showed that K5Nd crystal indexing was equally possible in both the trigonal and monoclinic space groups. The base structure of the K5Nd crystal can be described in a trigonal system, consisting of three distorted  $(\text{K},\text{Nd})\text{O}_6$  octahedra and six distorted  $\text{MoO}_4$  tetrahedra. K and Nd share the 3a Wyckoff position (site symmetry notation “ $-3m$ ”). The distance between the two nearest Nd atoms is approximately 5.98 Å. The monoclinic crystal system allows for an ordered distribution of K and Nd at the M positions of a palmierite-type structure.

The crystal structure of KNd5 belongs to the family of incommensurately modulated scheelite-type structures described within the (3+1)D superspace group  $I2/b(\alpha\beta 0)00$ . In this structure,  $\text{Nd}^{3+}$  ions are coordinated by eight oxygen atoms, forming  $\text{NdO}_8$  coordination polyhedra. The  $\text{K}^+$  and  $\text{Nd}^{3+}$  cations incommensurately share the 4e Wyckoff position located on a second-order rotation axis (site symmetry 2). The shortest Nd-Nd distance in the structure is approximately 4 Å.

## 4.2. Photoluminescence properties

### 4.2.1. Emission spectra

#### 4.2.1.1. Effect of $\text{RE}^{3+}$ substitution on emission characteristics

The PL spectra of K5Eu, K5Tb, and K5Nd crystals measured under 45 eV excitation to demonstrate the effect of  $\text{RE}^{3+}$  ion substitution on emission characteristics are depicted in Fig. 3. This high excitation energy corresponds to the interband transitions from the valence band or core levels to the conduction band,

allowing excitation of the whole set of emission centres in the crystals. The  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  crystal displayed emission lines in the 370–640 nm range, which resulted from the 4f-4f transitions in  $\text{Tb}^{3+}$ , see energy level scheme in Fig. 4. The weak blue emission lines in the 370–480 nm region were associated with transitions from the  $^5\text{D}_3$  level to the  $^7\text{F}_J$  ground states, whereas the more intense emissions in the 480–640 nm region originated from the  $^5\text{D}_4$  level to the  $^7\text{F}_J$  ( $J = 3–6$ ) levels. The most prominent emission was observed at 545 nm, corresponding to the  $^5\text{D}_4 - ^7\text{F}_5$  transition, responsible for the characteristic green luminescence of  $\text{Tb}^{3+}$ .

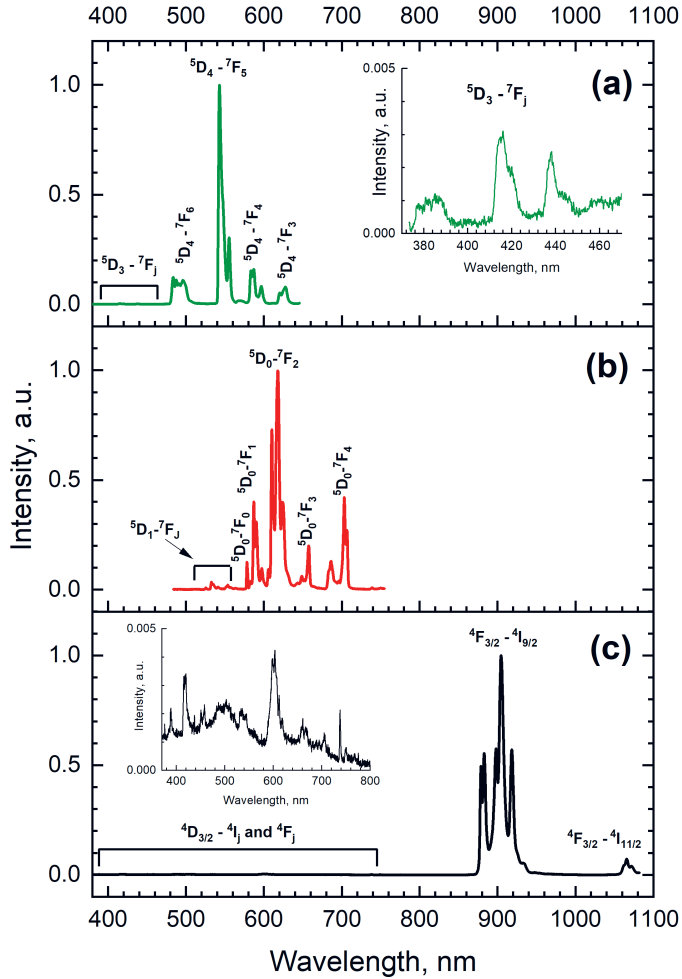


Figure 3: Luminescence spectra of  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  (a),  $\text{K}_5\text{Eu}(\text{MoO}_4)_4$  (b) and  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$  (c) measured under excitation 45 eV at  $T = 7$  K.

In contrast, the K5Eu crystal exhibited a series of sharp emission lines in the 520–720 nm range arising from intraconfigurational 4f–4f transitions of the  $\text{Eu}^{3+}$  ion. The emissions observed in the 580–720 nm region are attributed to transitions from the  $^5\text{D}_0$  excited state to the  $^7\text{F}_J$  ( $J = 0-4$ ) levels, with the most intense peak at 618 nm corresponding to the  $^5\text{D}_0 - ^7\text{F}_2$  transition, giving rise to strong orange-red luminescence. Relatively weaker bands in the 520–570 nm region originated from higher excited states  $^5\text{D}_1$  and  $^5\text{D}_2$ , which undergo partial radiative relaxation to the ground state before non-radiative decay to the  $^5\text{D}_0$  level.

In the  $\text{Nd}^{3+}$  substituted crystal (K5Nd), the emission characteristics change significantly, dominated by an intense near-infrared (NIR) emission arising from the  $^4\text{F}_{3/2} - ^4\text{I}_{9/2}$  and  $^4\text{F}_{3/2} - ^4\text{I}_{11/2}$  transitions within the  $4f^3$  configuration of  $\text{Nd}^{3+}$ . In addition to these strong NIR lines, weak but distinct emission lines are observed in the visible region, which are rarely reported for  $\text{Nd}^{3+}$  systems. These emissions are detectable under high excitation density, when higher-lying 4f levels become sufficiently populated. The observed visible emission lines are assigned to the  $^4\text{D}_{3/2} - ^4\text{I}_{11/2}$  (388 nm),  $^4\text{D}_{3/2} - ^4\text{I}_{13/2}$  (420 nm),  $^4\text{D}_{3/2} - ^4\text{I}_{15/2}$  (460 nm),  $^2\text{P}_{3/2} - ^4\text{I}_{15/2}$  (504 nm),  $^4\text{D}_{3/2} - ^4\text{F}_{3/2}$  (612 nm),  $^4\text{D}_{3/2} - ^4\text{F}_{5/2} + ^2\text{H}_{9/2}$  (655 nm),  $^4\text{D}_{3/2} - ^4\text{F}_{7/2} + ^4\text{S}_{3/2}$  (700 nm), and  $^2\text{P}_{3/2} - ^4\text{F}_{5/2} + ^2\text{H}_{9/2}$  (770 nm) transitions. Overall, the results clearly demonstrate that substituting different  $\text{RE}^{3+}$  ions into the  $\text{K}_5\text{RE}(\text{MoO}_4)_4$  crystal significantly shifts the emission wavelength, with  $\text{Tb}^{3+}$  yielding strong green emission,  $\text{Eu}^{3+}$  giving intense orange-red emission, and  $\text{Nd}^{3+}$  exhibiting dominant NIR emissions.

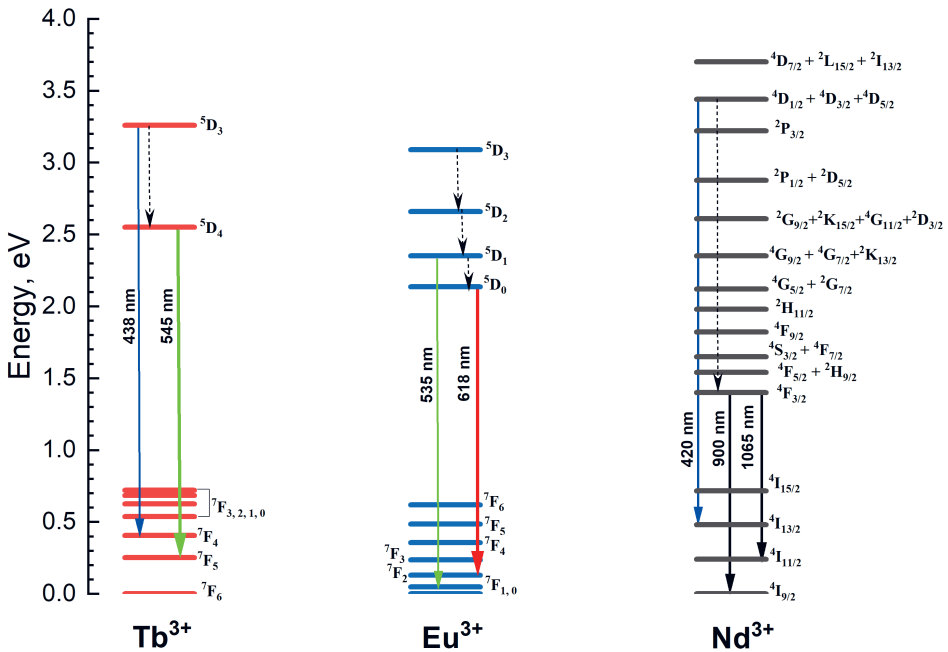


Figure 4: Energy level scheme of  $\text{Tb}^{3+}$ ,  $\text{Eu}^{3+}$ , and  $\text{Nd}^{3+}$ .

#### 4.2.1.2. Effect of structural variation on emission characteristics

Comparative studies of the photoluminescence of crystals with identical elemental composition but different crystal structures to understand the influence of structural variation on luminescence properties were performed on Tb<sup>3+</sup> and Nd<sup>3+</sup> based crystals. A pronounced dependence of the emission characteristics on crystal structure was observed, particularly between the K5Tb/KTb and K5Nd/KNd5 crystals.

The fine structure of the 4f electronic transitions was analysed in detail, allowing the observation of Stark splitting patterns associated with crystal-field interactions. Such measurements are particularly important because the spectral fine structure is highly sensitive to local symmetry, crystal-field strength, and structural distortions around the RE<sup>3+</sup> ions. Consequently, high-resolution spectroscopy provides valuable information regarding the occupation of crystallographic sites and the local coordination environment within the crystal lattice.

For the Tb<sup>3+</sup> based materials, the most intense emission in the 540–560 nm spectral range, corresponding to the <sup>5</sup>D<sub>4</sub>–<sup>7</sup>F<sub>5</sub> transition, was analysed in detail by recording high-resolution (0.4 nm) emission spectra for two potassium terbium molybdates exhibiting different crystal structures, KTb and K5Tb (Fig. 5). A noticeable difference in spectral line shape and fine structure of the emission spectra was observed for these two crystals. The ordered KTb crystal, in which Tb<sup>3+</sup> ions occupy TbO<sub>8</sub> polyhedra, exhibited a well-resolved fine structure consisting of 11 fine spectral lines. The appearance of the 11 resolved lines in KTb is consistent with the maximal crystal field splitting of the <sup>7</sup>F<sub>5</sub> multiplet into Stark levels, given by the 2J+1 rule in the case of a completely lifted degeneracy. The observation of the complete set of the emission lines in KTb therefore indicates a highly ordered local environment of Tb<sup>3+</sup> ions preferentially occupying one type of sites, in which the crystal field is sufficiently well defined to fully resolve the Stark splitting of the emitting level.

In the case of K5Tb, the emission lines were significantly smeared, suggesting a more complex, less uniform local crystal-field environment for the Tb<sup>3+</sup> ions. Structural analysis indicates that the Tb<sup>3+</sup> ions are coordinated by oxygen atoms in K5Tb, forming coordination polyhedra ranging from TbO<sub>6</sub> to TbO<sub>10</sub>. Such variability in coordination geometry leads to inhomogeneous broadening due to the overlapping of emission lines from multiple emitting centres.

Since multiple crystallographic sites may coexist in these materials, conventional luminescence spectra often consist of overlapping contributions from several emitting centres. Site-selective excitation enables probing of individual RE<sup>3+</sup> sites and thus provides a more detailed understanding of site-dependent emission behaviour.

To distinguish between the broadening arising from multiple crystallographic sites and that originating from structural disorder, the emission spectra were measured at 78 K by selecting the excitation energy in the 2.54–2.59 eV range, which corresponds to the broad line in excitation spectra arising due to the <sup>7</sup>F<sub>6</sub>–<sup>5</sup>D<sub>4</sub> Tb<sup>3+</sup> transitions (see Fig. 8). The results showed that the emission

profile remained consistently broadened without the emergence of additional resolvable site-specific features. This behaviour confirms that the observed spectral broadening in K5Tb is primarily determined by structural disorder rather than by the overlapping of emission lines originating from multiple non-equivalent Tb<sup>3+</sup> sites.

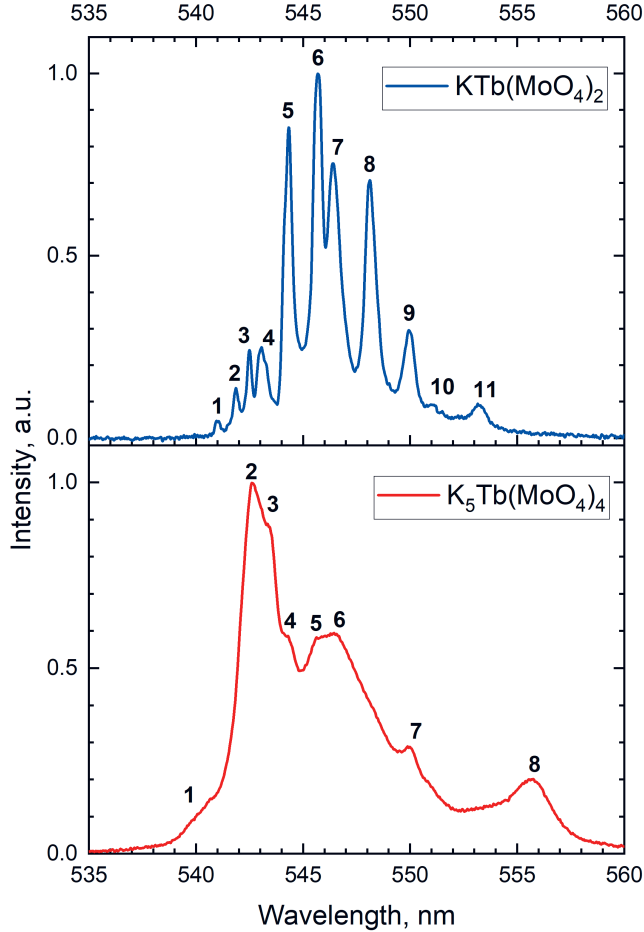


Figure 5: Luminescence spectra in the region of the  $^5\text{D}_4$ – $^7\text{F}_5$  transitions in KTb and K5Tb measured under excitation 2.55 eV, at  $T = 78$  K.

Similarly, crystal field splitting of the  $^4\text{I}_{9/2}$  state of  $\text{Nd}^{3+}$  ions in K5Nd and KNd5 showed pronounced differences, as depicted in Fig. 6. In K5Nd, the lines related to the transitions from the  $^4\text{F}_{3/2}$  state to different Stark components of the  $^4\text{I}_{9/2}$  state are observed within the 875–930 nm spectral range, whereas they form two distinct groups divided by a gap at about 890 nm. In KNd5, the corresponding group of lines is slightly shifted to the shorter wavelength range 860–930 nm, and the spectral positioning of the first four lines is more compact. Also, a distinct

redistribution of emission intensity between lines associated with different Stark levels is observed in the two crystals, which confirms that  $\text{Nd}^{3+}$  ions experience distinct local crystal-field environments in the two compounds, likely due to variations in site symmetry and coordination geometry in  $\text{NdO}_6$  octahedra and  $\text{NdO}_8$  polyhedra, hosting the  $\text{Nd}^{3+}$  ions in  $\text{K}_5\text{Nd}$  and  $\text{KNd}_5$ , respectively, which affect both energy splitting and transition probabilities.

The emission lines' width is similar in both crystals, indicating a similar level of structural disorder, contrary to the findings in potassium terbium molybdate crystals, where the emission lines of  $\text{Tb}^{3+}$  are substantially broadened in  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  compared to those in  $\text{KTb}(\text{MoO}_4)_2$  due to higher structural disorder in the former crystal [26].

Overall, these findings clearly demonstrate that  $\text{RE}^{3+}$  substitution and stoichiometric ratio driven structural changes noticeably influence the spectral characteristics of  $\text{RE}^{3+}$  emissions in double molybdates.

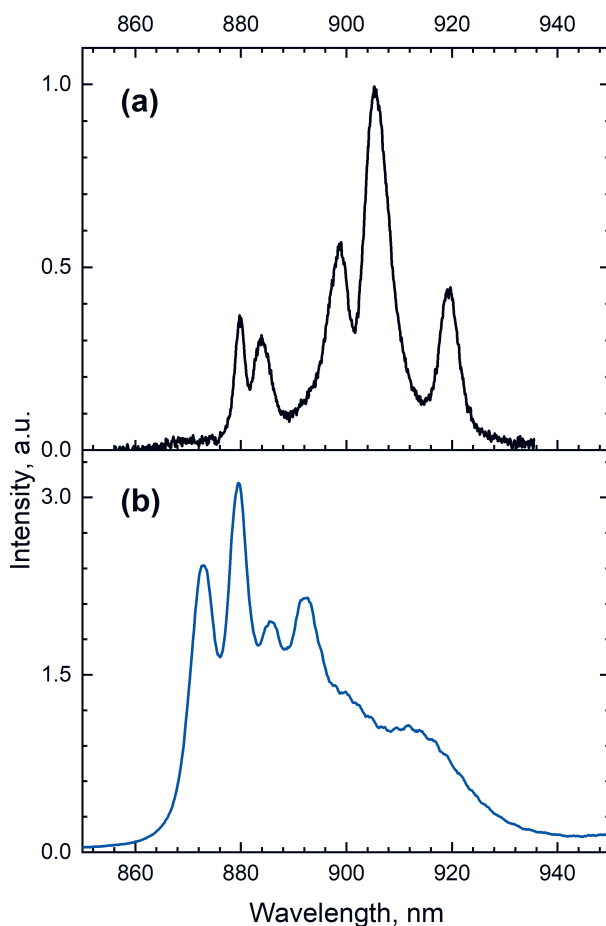


Figure 6. Luminescence spectra in the region of the  ${}^4\text{F}_{3/2} - {}^4\text{I}_{9/2}$  transitions in  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$  (a) and  $\text{KNd}_5(\text{MoO}_4)_8$  (b) measured under excitation 4.5 eV at  $T = 78$  K.

### 4.2.2. Photoluminescence excitation

Photoluminescence excitation spectra provide substantial information about the energy-level structure of materials involved in luminescence processes and are essential for selecting suitable excitation energies to investigate various relaxation mechanisms. To evaluate the influence of RE<sup>3+</sup> ion substitution and crystal structure on excitation characteristics, PLE spectra of all investigated crystals were measured over a wide excitation-energy range from 2.5 eV to deep VUV range (typically 45 eV) (Fig. 7). All studied crystals exhibit a broad excitation band in the 4–8 eV region, attributed to interband electronic transitions from the 2p O states at the top of the valence band to the 4d Mo states at the bottom of the conduction band. The presence of this broad band demonstrates efficient energy transfer from the MoO<sub>4</sub><sup>2-</sup> complex to the RE<sup>3+</sup> ions in all compounds. Since the broad excitation band appears at nearly the same spectral position in all K<sub>5</sub>RE(MoO<sub>4</sub>)<sub>4</sub> crystals, this indicates that the substitution of the RE<sup>3+</sup> ion does not significantly modify the electronic structure of the MoO<sub>4</sub><sup>2-</sup> complex within this crystal family. The bandgap variations are within 0.3 eV, depending on the composition. In contrast, substantial differences are observed in the crystal transparency region (2.5–4 eV), where narrow excitation lines corresponding to intra-configurational 4f–4f transitions of RE<sup>3+</sup> ions are present. Variations in the position, intensity, and number of these excitation lines arise from the different electronic configurations and 4f energy-level schemes of the individual rare-earth ions as well as crystal structure modifications.

To investigate the effect of crystal structure on excitation mechanisms, the PLE spectra of K5Tb were compared with those of KTb, and K5Nd with KNd5, over the same wide energy range (2.5–45eV). In both Tb<sup>3+</sup> and Nd<sup>3+</sup> based crystals, noticeable differences were observed in the spectral pattern, linewidth, and intensity of the 4f–4f excitation lines (Fig. 7). These variations are associated with differences in the local symmetry, coordination environment of the RE<sup>3+</sup> ions, and the degree of structural ordering resulting from different crystal stoichiometries.

A pronounced structural effect was observed for the Tb<sup>3+</sup> based crystals in the excitation lines associated with the <sup>7</sup>F<sub>6</sub>–<sup>5</sup>D<sub>4</sub> transitions of Tb<sup>3+</sup> (Fig. 8). In ordered KTb crystal, a well-resolved fine structure of the excitation lines is clearly visible, whereas in partially disordered K5Tb crystal, significant spectral broadening and smearing are observed. In addition to changes in the line shape, clear differences also appear in the position of the absorption edge and in the maximum of the broad excitation band, indicating the increase of the bandgap by 0.3 eV in K5Tb (Fig.7).

In the case of Nd<sup>3+</sup> based crystals, comparison of the PLE spectra of K5Nd and KNd5 (Fig. 7d and 7e) also revealed noticeable differences in the spectral fine structure and linewidth of the 4f–4f transitions. These differences similarly originate from variations in local symmetry and coordination environments around Nd<sup>3+</sup> ions in the two crystal structures. However, unlike the Tb based molybdates, the crystal structure does not appear to significantly influence the bandgap in Nd based compounds, as evidenced by the broad excitation bands occurring at nearly identical energies in both crystals and reflectivity spectra measurements.

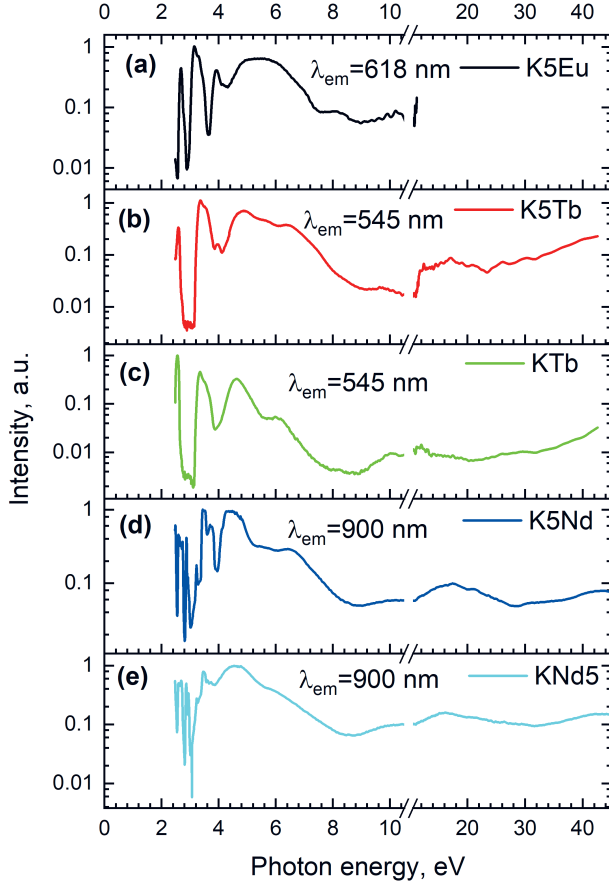


Figure 7: Photoluminescence excitation spectra of potassium rare earth double molybdates in semilogarithmic scale measured at 7 K.

The high-energy region of the excitation spectra provides further insight into energy-transfer processes involving free charge carriers. In particular, it is shown that the structural disorder influences the efficiency of energy transfer from the separated e-h pairs to  $\text{RE}^{3+}$ . At excitation energies above approximately 5.5 eV, typically separated electron-hole pairs are created, as previously reported for other alkali-earth molybdates [27]. In KTb, energy transfer from charge carriers to  $\text{Tb}^{3+}$  luminescence centres appears relatively inefficient, which is evidenced by the presence of only a weak shoulder near 6 eV (Fig. 7c). The intensity of this feature is nearly an order of magnitude lower than that of the main excitonic excitation band. In contrast, K5Tb exhibits much stronger broad excitation bands centred near 5.6 and 6.4 eV, whose intensities decrease only slightly relative to the excitonic band. These observations indicate considerably more efficient energy transfer from free carriers to the  $\text{Tb}^{3+} \ ^5\text{D}_4$  emitting states in the partially disordered lattice. Since such energy-transfer processes are commonly mediated by self-trapped excitons (STEs), structural disorder increases the probability of STE formation and thereby enhances the efficiency of energy transfer to  $\text{Tb}^{3+}$  centres.

In contrast to potassium terbium double molybdates, the excitation spectra of K5Nd and KNd5 demonstrate a pronounced decrease in intensity in the region of separated e–h pairs creation. This result also confirms the conclusion that despite their different crystal structures, these crystals are characterized by a similar rate of structural disorder.

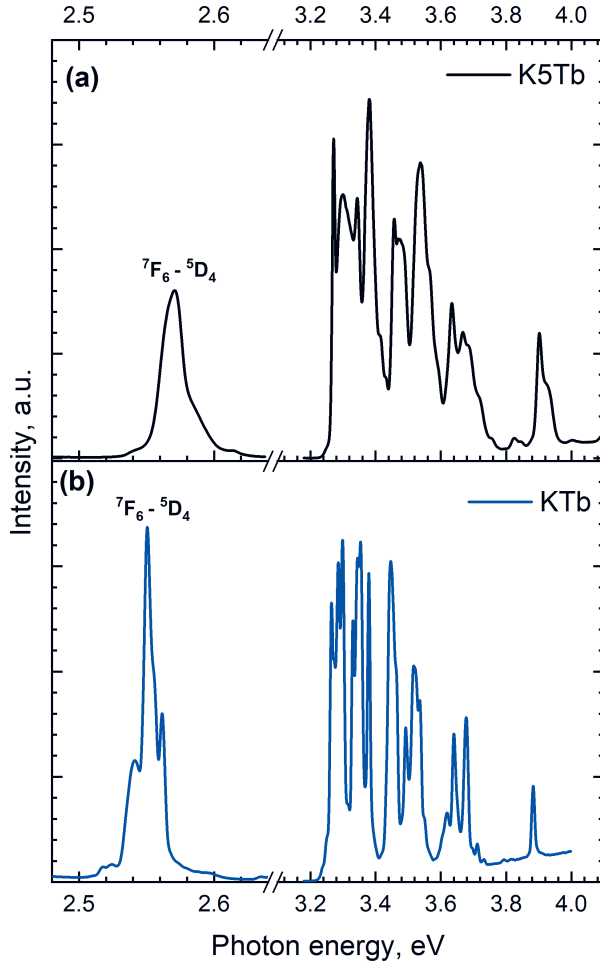


Figure 8: Photoluminescence excitation spectra of K5Tb (a) and KTb (b) measured with spectral resolution 0.4 nm,  $\lambda_{em}=545$  nm.

At higher excitation energies, approximately corresponding to  $2E_g$ , an increase in emission intensity was observed for all studied crystals. This enhancement is associated with the multiplication of electronic excitations (MEE) process, in which a single high-energy photon generates multiple electronic excitations. In addition to this conventional MEE process, Tb<sup>3+</sup> based crystals exhibited another type of MEE effect, regarded as a solid-state analogue of the Frank–Hertz effect. Under interband excitation, the onset of blue emission originating from the Tb<sup>3+</sup>  ${}^5D_3$  state was observed below the main MEE threshold, indicating that resonant

excitation of  $\text{Tb}^{3+}$  luminescence centres occurs through impact interaction between hot conduction electrons and  $\text{Tb}^{3+}$  ions. The experimentally observed threshold energies for the impact excitation of the blue emission agree well with the sum of the fundamental absorption edge energy and the excitation energy of the  $\text{Tb}^{3+} {}^5\text{D}_3$  level in both  $\text{KTb}$  and  $\text{K5Tb}$  crystals.

### 4.2.3. Luminescence decay kinetics

Luminescence decay kinetics measurements of the studied crystals provide information about the lifetimes of the emitting states and give insight into the radiative and non-radiative relaxation processes governing their emission behaviour. The investigated compounds, belonging to the  $\text{K}_5\text{RE}(\text{MoO}_4)_4$  structural type, exhibited decay times typical of the dominant emission of the respective  $\text{RE}^{3+}$  ions in other compounds. In particular,  $\text{K5Eu}$  shows red emission at 618 nm originating from the  $\text{Eu}^{3+} {}^5\text{D}_0 \rightarrow {}^7\text{F}_2$  transition, with a decay time of 1.420 ms. Similarly,  $\text{K5Tb}$  exhibits green emission at 545 nm from the  $\text{Tb}^{3+} {}^5\text{D}_4$  level, with a decay time of 1.583 ms, while  $\text{K5Nd}$  shows near-infrared emission at 1065 nm corresponding to the  $\text{Nd}^{3+} {}^4\text{F}_{3/2} \rightarrow {}^4\text{I}_{11/2}$  transition, with a decay time of 92  $\mu\text{s}$ .

The change of the crystal structure type of the compound formed by the same elements strongly influences the decay time. Significantly shorter decay times are observed in  $\text{KTb}$  (593  $\mu\text{s}$ ) and  $\text{KNd5}$  (6.65  $\mu\text{s}$ ) than those in  $\text{K5Tb}$  and  $\text{K5Nd}$ , respectively. This shortening of the decay time is associated with luminescence quenching caused by the higher concentration of  $\text{RE}^{3+}$  ions in these compounds. In these structures, the shorter RE-RE ion separation (approximately 4 Å compared to ~6 Å in the  $\text{K5RE}$  compounds) facilitates energy transfer between neighbouring rare-earth ions and promotes concentration quenching. The quenching process is primarily governed by non-radiative mechanisms such as cross-relaxation and Förster resonance energy transfer (FRET), both of which become increasingly efficient at shorter interionic distances.

In addition, a characteristic cascade relaxation mechanism within the  $\text{RE}^{3+} 4f$  electronic states is observed in all studied crystals except  $\text{KNd5}$ , manifested as rise components in the luminescence decay profiles. To illustrate this behaviour, the decay kinetics of  $\text{K5Eu}$  are discussed as a representative example (Fig. 9). The decay curve of the 618 nm emission was measured at 80 K under 395 nm (3.1 eV) excitation, corresponding to the  ${}^7\text{F}_0 \rightarrow {}^5\text{L}_6$  absorption transition. As shown in the inset of Fig. 9a, the decay profile exhibits a rise component followed by a single-exponential decay. Fitting with a two-exponential function yields a rise time of 11.0  $\mu\text{s}$  and a decay time of 1.420 ms. This rise time matched the decay time of the 535 nm emission originating from the  $\text{Eu}^{3+} {}^5\text{D}_1$  level (11.00  $\mu\text{s}$ , Fig. 9b), indicating that the population of the  ${}^5\text{D}_0$  emitting state is primarily governed by non-radiative relaxation from the higher-lying  ${}^5\text{D}_1$  state. Furthermore, an additional rise component of 0.550  $\mu\text{s}$  is observed in the  ${}^5\text{D}_1$  emission decay curve, which corresponds to feeding from the  ${}^5\text{D}_2$  level. This assignment is supported by independent decay measurements of the  ${}^5\text{D}_2$  emission (Fig. 9c), which exhibit two decay components of 0.40  $\mu\text{s}$  and 1.12  $\mu\text{s}$ , yielding an effective

rise time of approximately  $0.55 \mu\text{s}$  for the  ${}^5\text{D}_1$  emission. These results collectively confirm a stepwise, cascade-type relaxation process within the  $\text{Eu}^{3+}$  excited-state manifold, in which higher-lying 4f levels relax non-radiatively to lower states, ultimately populating the emitting  ${}^5\text{D}_0$  level responsible for the observed red luminescence at 618 nm.

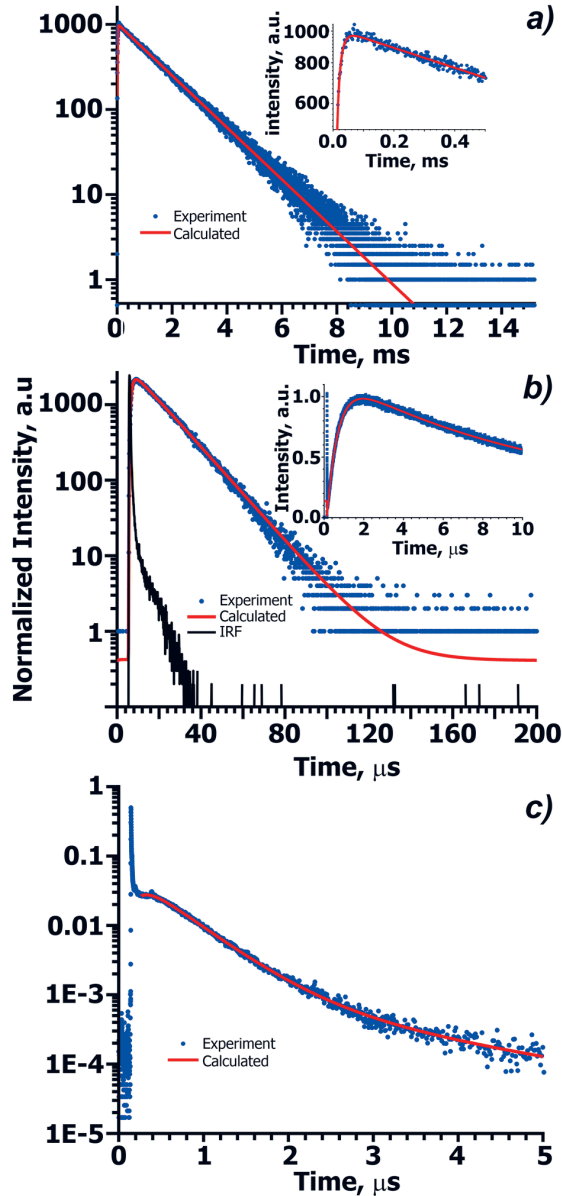


Figure 9. Decay curves of  $\text{K}_5\text{Eu}(\text{MoO}_4)_4$  emission measured for radiative transitions from  ${}^5\text{D}_0$  ( $\lambda_{\text{em}} = 618 \text{ nm}$ ) (a),  ${}^5\text{D}_1$  ( $\lambda_{\text{em}} = 535 \text{ nm}$ ) (b) and  ${}^5\text{D}_2$  ( $\lambda_{\text{em}} = 486 \text{ nm}$ ) (c) excited states of  $\text{Eu}^{3+}$ ,  $E_{\text{ex}} = 3.14 \text{ eV}$ ,  $T = 80 \text{ K}$  for  ${}^5\text{D}_0$  and  ${}^5\text{D}_1$ , and RT for  ${}^5\text{D}_2$ . In the inset: initial part of the decay, measured with 3-ns pulses of an OPO laser at  $E_{\text{ex}} = 3.14 \text{ eV}$ .

A similar cascade relaxation mechanism has been observed in K5Tb, KTb, and K5Nd. In Tb based crystals, the electrons relax from the  $^5D_3$  to  $^5D_4$  state of  $Tb^{3+}$  ions. Similarly, in K5Nd, the relaxation undergoes from the  $^4D_{3/2}$  to  $^4F_{3/2}$  state of  $Nd^{3+}$  ions. However, such processes are not clearly resolved in K5Nd due to the extremely short luminescence lifetime (less than 1 ns) of the higher-lying excited state from which electrons non-radiatively relax to the lowest excited state.

#### 4.2.4. Influence of temperature on luminescence of the double molybdates

Temperature-dependent luminescence studies were conducted to investigate thermal stability of  $RE^{3+}$  luminescence in double molybdates and to clarify how it is affected by the variation of crystal structure and excitation energy. Significant differences in the emission quenching temperature ( $T_q$ ) were observed in the  $K_5RE(MoO_4)_4$  system containing  $Eu^{3+}$ ,  $Tb^{3+}$  or  $Nd^{3+}$ , where  $T_q$  is defined as the temperature at which emission intensity decreases by a factor of two. These differences arise from different electronic configurations of the rare-earth ions, which result in different 4f energy-level structures and varying strengths of electron-phonon interaction. Consequently, the thermal stability of emission varies considerably across the studied systems. Among the investigated compounds, K5Eu exhibits the highest thermal stability, showing the quenching temperature  $T_q = 564$  K for the 618 nm emission under excitation from the ground  $^7F_0$  level to the  $^5D_2$  level at 2.66 eV (Fig. 10). In comparison, K5Tb shows  $T_q = 380$  K for the  $^7F_6-^5D_4$  transition (Fig. 11a, left panel), while K5Nd exhibits a significantly lower value of  $T_q = 200$  K for the  $^4F_{3/2}-^4I_{9/2}$  transition [28]. In addition, both K5Eu and K5Tb systems showed a systematic decrease in quenching temperature with increasing excitation energy.

In K5Eu,  $T_q$  decreases from 564 K for the  $^7F_0-^5D_2$  transition at 2.66 eV to 475 K for the  $^7F_0-^5L_6$  transition at 3.14 eV, and further to 434 K for the  $^7F_0-^5D_4$  transition at 3.41 eV (Fig. 10). For the higher-energy  $^7F_0-^5H_1$  transition at 3.87 eV, the onset of thermal quenching occurred below the low-temperature limit of our setup, and the temperature dependence became similar to that observed under excitation in the fundamental absorption region. The excitation at 5 eV (Fig. 10) corresponds to the creation energy of excitons possessing low thermal stability in K5Eu. A similar trend is observed for the green emission of K5Tb, where  $T_q$  decreases from 380 K for the  $^7F_6-^5D_4$  transitions to 330 K for  $^7F_6-^5D_3$ ,  $^5G_6$  and further to 267 K for the  $^7F_6-^5D_0$  transitions (Fig. 11a, b, and c1(left panel)). Under interband excitation,  $T_q$  decreases to approximately 120 K (Fig. 11 c2), indicating inefficient exciton-mediated energy transfer at elevated temperatures, as also observed for K5Eu.

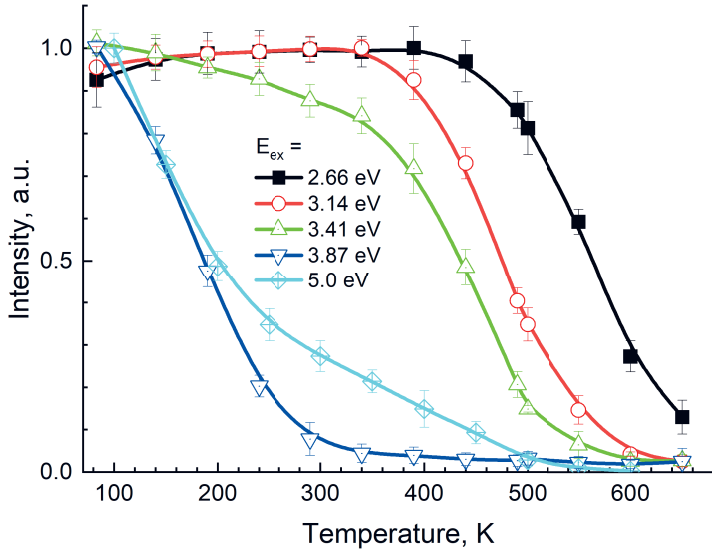


Figure 10: Temperature dependences of the  $\text{Eu}^{3+}$  emission ( $\lambda_{\text{em}}=620$  nm) in K5Eu for intracentre ( $E_{\text{ex}}=2.66, 3.14, 3.41,$  and  $3.87$  eV) and interband ( $E_{\text{ex}}=5$  eV) transitions from the  ${}^7\text{F}_0$  ground state of  $\text{Eu}^{3+}$ .

The observed shift in  $T_q$  with excitation energy in K5Eu and K5Tb can be attributed to the overlap between the  $\text{RE}^{3+}$  4f-4f absorption lines and the Urbach tail of the fundamental absorption edge. The Urbach tail shifts to lower energies with increasing temperature due to enhanced electron-phonon interaction, leading to a temperature-dependent increase in absorption in the relevant spectral region. The absorption coefficient in the Urbach tail can reach  $10^5$ – $10^6$   $\text{cm}^{-1}$ , which is two to three orders of magnitude higher than that of parity-forbidden 4f-4f transitions. As this overlap increases, excitation is progressively dominated by interband transitions and exciton formation. However, due to relatively low thermal stability of excitons transferring their energy to  $\text{RE}^{3+}$  ions, this results in enhanced thermal quenching of  $\text{RE}^{3+}$  emission.

In contrast, K5Nd exhibits nearly identical temperature dependences under intracentre excitation at 3.5 eV and direct exciton creation at 4.5 eV. This indicates highly efficient, temperature-independent energy transfer from excitons to  $\text{Nd}^{3+}$  centres up to at least 120 K. The similarity of these dependences suggests strong exciton localization and relatively high thermal stability of excitons in the Nd based system.

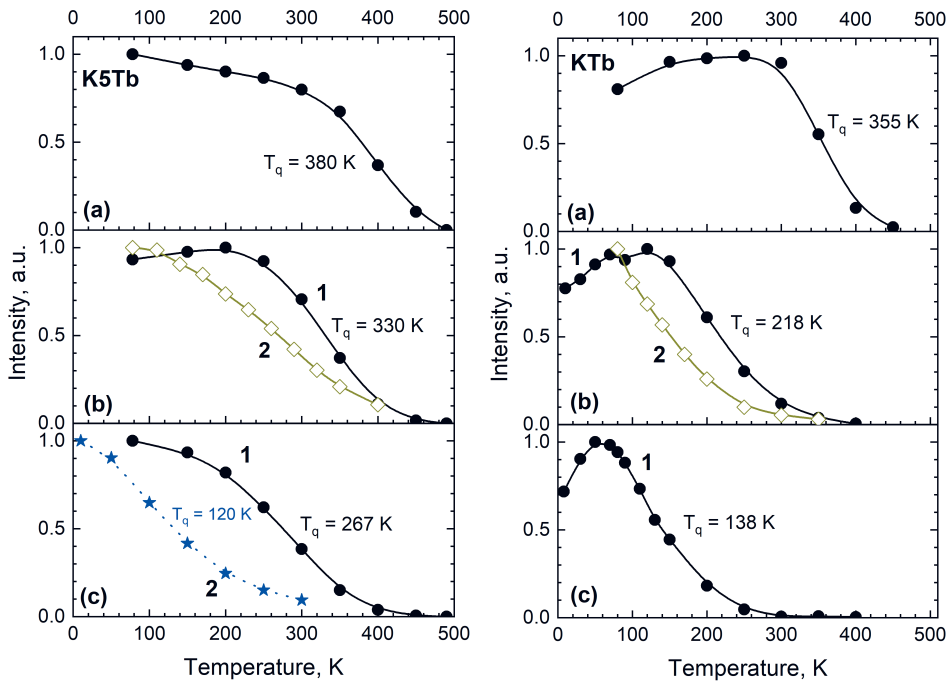


Figure 11. Temperature dependences of the  $\text{Tb}^{3+}$  545-nm emission intensity for the transitions  ${}^7\text{F}_6\text{-}{}^5\text{D}_4$  (a),  ${}^7\text{F}_6\text{-}{}^5\text{D}_3$  (b),  ${}^7\text{F}_6\text{-}{}^5\text{D}_0$  (c1), and interband transitions (4.86 eV, c2) in K5Tb (left panel) and KTb (right panel). Curves b2 represent the temperature dependence of the  $\text{Tb}^{3+}$  438 nm emission intensity under the  ${}^7\text{F}_6\text{-}{}^5\text{D}_3$  transition (3.28 eV) excitation energy.

The influence of crystal structure on thermal stability of luminescence was analysed by comparing K5Tb with KTb and K5Nd with KNd5. In the Tb-based system, pronounced differences in thermal stability were observed. In KTb, the 545 nm emission excited in the region of the  ${}^7\text{F}_6\text{-}{}^5\text{D}_4$  transition has a lower  $T_q$  of approximately 355 K, compared to 380 K observed in K5Tb. For the higher-energy excitations corresponding to the  ${}^7\text{F}_6\text{-}{}^5\text{D}_3$ ,  ${}^5\text{G}_6$ , and  ${}^7\text{F}_6\text{-}{}^5\text{D}_0$  transitions,  $T_q$  decreases further to about 218 K and 138 K, respectively, these values also being lower than in K5Tb showing  $T_q$  equal to 330 K and 267 K for these transitions. These differences are primarily due to different band-gap energies in KTb and K5Tb. The narrower band gap in KTb leads to the overlap between the  $\text{Tb}^{3+}$  4f states and the Urbach tail of the absorption edge at lower temperatures. Consequently, thermal quenching occurs at lower temperatures in KTb. Additionally, thermal ionization processes may contribute, as higher excited 4f states lie closer to the conduction band.

A similar distinction is observed in the blue emission at 438 nm related to the  ${}^5\text{D}_3\text{-}{}^7\text{F}_4$  transition. In KTb, no clear thermal stability plateau is observed even at 80 K, indicating an early onset of the luminescence quenching. In contrast, K5Tb showed higher luminescence thermal stability with  $T_q=265$  K. This difference is explained by the higher energy position of the conduction band in K5Tb, which delayed overlap with  $\text{Tb}^{3+}$  excited states.

In contrast, a negligible effect of crystal structure on the thermal stability of NIR luminescence was observed in the  $\text{Nd}^{3+}$  based systems. Temperature-dependent emission measurements performed under both intracentre ( $E_{\text{exc}} = 3.5 \text{ eV}$ ) and inter-band excitation ( $E_{\text{exc}} = 4.5 \text{ eV}$ ) revealed a comparable decrease in emission intensity with increasing temperature for both K5Nd and KNd. These results indicate that the thermal quenching mechanisms are similar in both compounds [28].

## SUMMARY

This work is devoted to the investigation of the influence of elemental and structural modifications including structural disorder in potassium rare earth double molybdates on the electronic band structure, energy transfer efficiency to the emission centres and their luminescence characteristics. A set of single crystals of  $\text{K}_5\text{Eu}(\text{MoO}_4)_4$ ,  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$ ,  $\text{KTb}(\text{MoO}_4)_2$ ,  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$ , and  $\text{KNd}_5(\text{MoO}_4)_8$  were studied using a variety of experimental methods including time-resolved photoluminescence spectroscopy using synchrotron radiation with the focus placed on the dependence of energy transfer processes to  $\text{RE}^{3+}$  emission centres on temperature. The performed study highlights the importance of crystal structure type and ordering in determining the photoluminescence characteristics of  $\text{RE}^{3+}$  ions and its thermal stability in double molybdates.

The main conclusions are the following:

1. The emission typical of the intra-configurational 4f–4f transitions in  $\text{RE}^{3+}$  ions was observed in luminescence spectra. The substitution of  $\text{RE}^{3+}$  ions enables emission colour tuning in a broad spectral range, with  $\text{Tb}^{3+}$  producing intense green emission 480–640 nm corresponding to the  $^5\text{D}_4$ – $^7\text{F}_j$  transitions,  $\text{Eu}^{3+}$  exhibiting red emission in the 520–720 nm region due to the  $^5\text{D}_0$ – $^7\text{F}_j$  transitions, and  $\text{Nd}^{3+}$  emitting in the near-infrared region at 850–950 nm due to the  $^4\text{F}_{3/2}$ – $^4\text{I}_{9/2}$ ,  $^4\text{I}_{11/2}$  transitions.
2. For the first time intrinsic emission of self-trapped excitons was detected in double potassium molybdates  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$ , and  $\text{KNd}_5(\text{MoO}_4)_8$ . An unusually short lifetime of the STE emission in the nanosecond range and its low intensity are ascribed to a rapid energy transfer from the STEs to the  $\text{Nd}^{3+}$  ions.
3. It is shown that the crystal structure exerts a profound influence on the luminescence characteristics. The incommensurately modulated  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  crystal exhibited a broader and more intense green emission than the ordered  $\text{KTb}(\text{MoO}_4)_2$  crystal, which is ascribed to the availability of multiple non-equivalent sites in the lattice and the effect of partial structural disorder of crystal structure of  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$ . The luminescence intensity enhancement in  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  is attributed to the structural disorder induced suppression of concentration quenching, as a result of the hindering of nonradiative energy transfer between neighbouring  $\text{Tb}^{3+}$  ions.
4. The studies of the reflectivity and excitation spectra indicated a negligible contribution of the electronic states of potassium and rare-earth ions to the formation of the valence band and the bottom of the conduction band in the investigated compounds. In particular, the analysis of reflection spectra of  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$  and  $\text{KNd}_5(\text{MoO}_4)_8$  demonstrated that the main features observed in the fundamental absorption region can be attributed to the electronic transitions from the 2p O states forming the valence band to the 4d Mo states of the conduction band. The contribution of potassium electronic states is

shown to be remarkable for  $\text{K}_5\text{Nd}(\text{MoO}_4)_4$  in the region 18–28 eV, which is ascribed to the electron transitions from the 3p K core states to high-density states in the conduction band as well as to the formation of core excitons with a hole component localized at the 3p K states.

5. The analysis of the excitation spectra in the fundamental absorption region allowed to clarify the peculiarities of the energy transfer from separated electrons and holes to  $\text{RE}^{3+}$  ions. The absence of the UV charge-transfer band in  $\text{K}_5\text{Eu}(\text{MoO}_4)_4$  and inter-valence charge-transfer band below the fundamental absorption edge of  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  and  $\text{KTb}(\text{MoO}_4)_2$  indicated that the ground 4f levels of  $\text{RE}^{3+}$  are located within or near the valence band in double molybdates. It is concluded that energy transfer from separated electrons and holes in the conduction and valence bands, respectively, is realized via an intermediate self-trapped exciton state.
6. The efficiency of the energy transfer from separated electrons and holes to  $\text{RE}^{3+}$  emission centres is shown to depend on the disorder rate of the crystal structure. It is established that the structural disorder promotes exciton formation by decreasing the mean free path length of charge carriers during the migration stage.
7. For  $\text{K}_5\text{Tb}(\text{MoO}_4)_4$  and  $\text{KTb}(\text{MoO}_4)_2$  crystals, different mechanisms are proposed for energy transfer to the excited  ${}^5\text{D}_4$  and  ${}^5\text{D}_3$  terms, which are responsible for the green and blue sets of emission lines. The energy transfer from STEs to  $\text{Tb}^{3+}$  is realized via the excitation of the lowest excited term  ${}^5\text{D}_4$ , giving rise only to the green emission. The term  ${}^5\text{D}_3$  responsible for the blue emission can be excited either via the intra-centre excitation of  $\text{Tb}^{3+}$  or via the resonant excitation of the  $\text{Tb}^{3+}$  ions through the inelastic relaxation of hot electrons in the conduction band (so called solid-state analogue of the Frank-Hertz effect).
8. It is established that the thermal stability of the  $\text{RE}^{3+}$  emission depends strongly on excitation energy. In the  $\text{Eu}^{3+}$  and  $\text{Tb}^{3+}$  based systems, the quenching temperature decreases with increasing excitation energy due to the enhanced overlap of the  $\text{RE}^{3+}$  4f excited states with the temperature-dependent Urbach tail of the fundamental absorption edge. Under the interband excitation above the fundamental absorption edge, thermal quenching occurs at significantly lower temperatures determined by the stability of the STE state. Crystal structure variation significantly influenced the luminescence thermal stability in Tb-based double molybdates, with the bandgap varying with structure, thereby defining the rate of thermal ionization of the emission centre. In general, the variation of thermal stability of the emission with the excitation energy is attractive for the application of these crystals in non-contact luminescence thermometry.

9. The decay kinetics measurements showed that the  $K_5RE(MoO_4)_4$  compounds exhibit long luminescence lifetimes ranging from milliseconds for  $K_5Eu(MoO_4)_4$ , and  $K_5Tb(MoO_4)_4$  to hundred microseconds for  $K_5Nd(MoO_4)_4$ . In contrast,  $KTb(MoO_4)_2$  and  $KNd_5(MoO_4)_8$  show significantly shorter decay times due to concentration quenching arising from the reduced RE-RE distances and enhanced nonradiative cross-relaxation and Förster resonance energy transfer. A cascade type relaxation between various excited states of  $RE^{3+}$  is observed in all investigated double molybdates, except  $KNd_5(MoO_4)_8$ , in which this relaxation is too fast to be resolved by the experimental setups used.

# KOKKUVÕTE

## Energia ülekanne kiirgustsentritele leelismetallide ja haruldaste muldmetallide molübdaatides

Doktoritöö on pühendatud elementkoostise ja struktuursete muutuste, sealhulgas struktuurse korratuse, mõju uurimisele kaaliumi ja haruldaste muldmetallide topeltmolübdaatide elektroonse tsoonstruktuurile, energia ülekande efektiivsusele kiirgustsentritele ning luminesentsi omadustele. Uuriti  $K_5Eu(MoO_4)_4$ ,  $K_5Tb(MoO_4)_4$ ,  $KTb(MoO_4)_2$ ,  $K_5Nd(MoO_4)_4$ , ja  $KNd_5(MoO_4)_8$  monokristalle, kasutades erinevaid eksperimentaalseid meetodeid, sealhulgas aeglahutusega fotoluminesents-spektroskoopiat kasutades sünkrotronkiirgust ning pöörates erilist tähelepanu  $RE^{3+}$  kiirgustsentritele energia ülekande protsesside temperatuurisõltuvusele. Läbiviidud uuring rõhutab kristallstruktuuri tüübi ja korratuse olulist rolli haruldaste muldmetallide ionide fotoluminesentsi omaduste ning selle termilise stabiilsuse määramisel topeltmolübdaatides.

Peamised järeldused on järgmised:

1. Luminesents-spektrites täheldati  $RE^{3+}$  ionide iseloomulikke  $4f-4f$  sisekonfiguratsiooniliste üleminekutele vastavaid kiirguseid.  $RE^{3+}$  ionide asendamine võimaldab kiirguse värvi häälestamist laias spektraalses vahemikus:  $Tb^{3+}$  tekitab intensiivset rohelist kiirgust 480–640 nm vahemikus, mis vastab üleminekutele  $^5D_4-^7F_J$ ;  $Eu^{3+}$  kiirgab punases piirkonnas 520–720 nm  $^5D_0-^7F_J$  üleminekute tõttu;  $Nd^{3+}$  kiirgab lähisinfra-punases piirkonnas 850–950 nm, mis vastab  $^4F_{3/2}-^4I_{9/2}$  ja  $^4I_{11/2}$  üleminekutele.
2. Esmakordselt tuvastati topeltmolübdaatides  $K_5Nd(MoO_4)_4$  ja  $KNd_5(MoO_4)_8$  iselöksustunud eksitonidega seotud omakiirgus. Seda kiirgust iseloomustab ebataavaliselt lühike eluiga nanosekundite ajavahemikus, ning selle madal intensiivsus seostati kiire energia ülekandega eksitonidelt  $Nd^{3+}$  ionidele.
3. Näidati, et kristallstruktuur mõjutab oluliselt luminesentsi omadusi. Kõrglahutusega ergastusspektrite analüüs näitas, et  $Eu^{3+}$  ionid asuvad  $K_5Eu(MoO_4)_4$ -s vähemalt kolmes mitteekvivalentses kristallograafilises positsioonis, mis on seotud selle kristalli moduleeritud superstruktuuriga. Superstruktuurset moduleeritud  $K_5Tb(MoO_4)_4$  kristallil on laiem ja intensiivsem roheline kiirgusriba võrreldes korrastatud  $KTb(MoO_4)_2$  kristalliga, mida seotatakse mitmete mitteekvivalentsete  $Tb^{3+}$  sõlmede olemasoluga võres ning osalise struktuurse korratuse mõjuga. Luminesentsi intensiivsuse suuremine  $K_5Tb(MoO_4)_4$ -s on seotud struktuurset korratusest tingitud kontsentratsioonilise kustumise pärssimisega läbi mittekiirguslikku energiaülekande takistamise lähedal asuvate  $Tb^{3+}$  ionide vahel.
4. Elektroonilise tsoonstruktuuri uuringud näitasid, et kaaliumi ja haruldaste muldmetallide ionide elektroonsetel olekul on uuritud  $K_5RE(MoO_4)_4$  ühendites tühine mõju valents- ja juhtivustsoonide kujunemisele. Eelkõige näitas

$K_5Nd(MoO_4)_4$  ja  $KNd_5(MoO_4)_8$  peegeldusspektrite analüüs, et põhilised tunnused, mis on jälgitavad fundamentaalneeldumise piirkonnas, tulenevad üleminekutest 2p O seisunditelt valentstsoonis 4d Mo seisunditele juhtivustsoonis. Kaaliumi elektroonsete olekute panus on oluline 18–28 eV piirkonnas, mida seostatakse üleminekutega 3p K sisekihilt juhtivustsooni kõrge tihedusega elektronseisunditele ning süvakihieksitoni tekkega, mille augukomponent on lokaliseerunud 3p K seisundis.

5. Ergastusspektrite analüüs fundamentaalneeldumispirkonnas võimaldas selgitada välja energia ülekande iseärasusi eraldatud elektronidelt ja aukudelt  $RE^{3+}$  ioonidele. Laenguülekande UV riba puudumine  $K_5Eu(MoO_4)_4$  kristallis ning intervalentse laenguülekande riba puudumine  $K_5Tb(MoO_4)_4$  ja  $KTb(MoO_4)_2$  puhul fundamentaalneeldumise ääre all näitab, et  $RE^{3+}$  ioonide 4f põhi-seisundid asuvad topeltmolübdfaatides valentstsooni sees. Järeldatakse, et energia ülekande eraldatud elektronidelt ja aukudelt toimub vahepealse iselõksustunud eksitoni seisundi kaudu.
6. Näidati, et energia ülekande efektiivsus eraldatud elektronidelt ja aukudelt  $RE^{3+}$  kiirgustsentriale sõltub kristallstruktuuri korratuse määrast. Leiti, et struktuurne korratus soodustab eksitonide teket, vähendades laengukandjate keskmist vabatee pikkust migratsioonifaasis.
7.  $K_5Tb(MoO_4)_4$  ja  $KTb(MoO_4)_2$  kristallide puhul pakuti välja erinevaid energia ülekande mehhanisme ergastatud  ${}^5D_4$  ja  ${}^5D_3$  nivooale, mis vastutavad roheline ja sinise kiirguse eest. Energiaülekande iselõksustunud eksitonidelt  $Tb^{3+}$  ioonidele toimub madalaimale  ${}^5D_4$  ergastatud seisundile, mis annab ainult rohelist kiirgust.  ${}^5D_3$  seisund, mis vastutab sinise kiirguse eest, võib ergastuda kas  $Tb^{3+}$  tsentrisese ergastuse kaudu või resonantse ergastuse kaudu juhtivustsooni kuumade elektronide mitteelastse relaksatsiooni käigus (nn Frank–Hertzi efekti tahkisanaloog).
8. Leiti, et  $RE^{3+}$  kiirguse termiline stabiilsuse määrab tugevalt nii haruldaste muldmetallide asendamine kui ka ergastuse energia.  $K_5RE(MoO_4)_4$  ühendite seas näitab  $K_5Eu(MoO_4)_4$  kõrgeimat termilise kustumise temperatuuri  $T_q$ , millele järgnevad  $K_5Tb(MoO_4)_4$  ja  $K_5Nd(MoO_4)_4$ .  $Eu^{3+}$  ja  $Tb^{3+}$  süsteemides väheneb  $T_q$  samuti ergastuse energia suurenedes, kuna  $RE^{3+}$  4f ergastatud seisundite kattuvus temperatuurist sõltuva Urbachi sabaga fundamentaalneeldumise ääre lähedal suureneb. Tsoonidevahelise ergastuse korral fundamentaalneeldumise äärest kõrgemal toimub termiline kustumine oluliselt madalamatel temperatuuridel, mille määrab eksitonseisundi madal stabiilsus. Kristallstruktuuri varieerumine mõjutab oluliselt kiirguse termilist stabiilsust terbiumil põhinevates topeltmolübdfaatides, kus keelutsooni laius sõltub kristallstruktuurist, mis määrab kiirgustsentri termilise ionisatsiooni kiiruse. Üldiselt on kiirguse termilise stabiilsuse sõltuvus ergastuse energiast atraktiivne nende kristallide kasutamiseks kontaktivabas luminescentsents-termomeetrias.

9. Kiirguse kustumiskineetika mõõtmised näitasid, et  $K_5RE(MoO_4)_4$  ühenditel on luminesentsi eluajad pikad, ulatudes millisekunditest  $K_5Eu(MoO_4)_4$  ja  $K_5Tb(MoO_4)_4$  puhul kuni sadade mikrosekunditeni  $K_5Nd(MoO_4)_4$  puhul. Seevastu  $KTb(MoO_4)_2$  ja  $KNd_5(MoO_4)_8$  näitavad oluliselt lühemaid kustumisaegu, mis on tingitud kontsentratsioonilisest kustumisest, mis on tingitud väiksematest RE-RE kaugustest ning vastavalt efektiivsemast mittekiirguslikust ristrelaksatsioonist ja Försteri resonantsest energiaülekandest. Kaskaadrelaksatsiooni erinevate ergastatud  $RE^{3+}$  seisundite vahel täheldati kõigis uuritud topeltnolüüdaatides, välja arvatud  $KNd_5(MoO_4)_8$ , kus see relaksatsioon on liiga kiire, et selle aega kasutatud eksperimentaalse seadmete abil määrata.

## ACKNOWLEDGEMENTS

I would like to express my sincere gratitude to my supervisors, Dr. Vitali Nagirnyi and Dr. Dmitry Spassky, for their continuous support, guidance, and encouragement throughout my doctoral studies and the preparation of this thesis. I am also grateful to the Head of the Laboratory of Physics of Ionic Crystals, Prof. Aleksandr Lushchik, for providing the opportunity to conduct my research in this laboratory.

I sincerely thank Prof. Marco Kirm, Dr. Eduard Feldbach, and Dr. Ivo Romet for their expertise in spectroscopy and valuable technical assistance during the experimental work. My gratitude is extended to B. Redkin and V. A. Morozov for providing the crystals studied in this work, as well as to Dr. Hugo Mändar for X-ray diffraction (XRD) analysis and to Mr. Peeter Ritslaid for X-ray fluorescence (XRF) measurements from the Laboratory of Thin Films. I also acknowledge Dr. Kirill Chernenko for his support during experiments at the FinEstBeAMS beamline (MAX IV Laboratory, Lund) and Dr. Aleksei Kotlov and Dr. Yevheniia Smortsova for their assistance at the P66 beamline at DESY. I am also thankful to all members of the Laboratory of Physics of Ionic Crystals for their collegial atmosphere, support, and discussions throughout my studies.

Finally, I express my deepest gratitude to my family and friends for their unwavering support and encouragement during my academic journey.

This work has been supported by Estonian Research Council grant PRG2733, the ERDF funding in Estonia granted to the Centre of Excellence TK210 “Centre of Excellence in Sustainable Green Hydrogen and Energy Technologies” and TWINNING project “EXANST” funded by the European Union under grant agreement No. 101159716.

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## **PUBLICATIONS**

## CURRICULUM VITAE

**Name:** Muhammad Usama Jamal  
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### Education:

2020–2022 MSc, Materials Science and Technology, University of Tartu, Estonia  
2022–2026 PhD, University of Tartu

### Languages:

Urdu Native  
English Good  
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### Career:

2022–2026 Junior research fellow, Laboratory of Physics of Ionic Crystals, Institute of Physics, University of Tartu, Estonia.

### Research interests:

Solid-state physics, crystal growth and materials synthesis, luminescent materials (phosphors and scintillators), and steady-state and time-resolved optical spectroscopy.

### Publications:

- M.U. Jamal**, V. Nagirnyi, H. Mändar, K. Chernenko, D. Spassky, Structural and luminescence properties of  $K_5Nd(MoO_4)_4$  and  $KNd_5(MoO_4)_8$ , *Journal of Luminescence* (2026) 121995. <https://doi.org/10.1016/j.jlumin.2026.121995>.
- M.U. Jamal**, V. Nagirnyi, K. Chernenko, A. Kotlov, Y. Smortsova, D. Spassky, Crystal structure controlled energy transfer to  $Tb_{3+}$  in  $KTb(MoO_4)_2$  and  $K_5Tb(MoO_4)_4$  crystals, *Materials Research Bulletin* 191 (2025) 113553. <https://doi.org/10.1016/j.materresbull.2025.113553>.
- D. Spassky, A. Vasil'ev, **M.U. Jamal**, V.A. Morozov, B.I. Lazoryak, B.S. Redkin, K. Chernenko, V. Nagirnyi, Temperature dependent energy transfer to  $Eu^{3+}$  emission centres in  $K_5Eu(MoO_4)_4$  crystals, *CrystEngComm* 26 (2024) 1106–1116. <https://doi.org/10.1039/D3CE01201H>.

### **Conference presentations:**

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry, “Influence of crystal structure on energy transfer to  $Tb^{3+}$  in  $K_5Tb(MoO_4)_4$  and  $KTb(MoO_4)_2$  crystals”, oral presentation at the FMNT & NIBS Conference, October 2024, Tartu, Estonia.

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry, “The influence of crystal structure on luminescence properties of  $Tb^{3+}$  in double molybdates”, poster presentation at the 8th Baltic electrochemistry conference, April 2024.

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry, “The impact of crystal structure on  $Tb^{3+}$  luminescence in double molybdates”, poster presentation at the LUMDETR 2024 Conference, June 2024, Riga, Latvia.

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry; “Luminescence properties of  $K_5Eu(MoO_4)_4$ ”, poster and 3-minute oral presentation at the GSFMT Scientific Conference, 2023, Tartu, Estonia.

### **Awards and scholarships:**

2024 Received mobility grant from Estonian Doctoral School to visit the LUMDETR 2024 Conference, Riga, Latvia.

2020–2022 Awarded specialization and performance scholarships multiple times during my master’s studies.

## ELULOOKIRJELDUS

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### Haridus:

2020–2022 MSc, materjaliteadus ja tehnoloogia, Tartu Ülikool, Eesti  
2022–2026 Doktoriõpe, Tartu Ülikool

### Keelteoskus:

Urdu keel emakeel  
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2022–2026 Nooremteadur, Ioonkristallide füüsika labor, Füüsika Instituut, Tartu Ülikool.

### Peamised uurimisvaldkonnad:

Tahkisfüüsika, kristallide kasvatamine ja materjalide süntees, luminescentsmaterjalid (fosfoorid ja stsintillaatorid) ning statsionaarne ja ajalise lahutusega optiline spektroskoopia.

### Publikatsioonid:

- M.U. Jamal**, V. Nagirnyi, H. Mändar, K. Chernenko, D. Spassky, Structural and luminescence properties of  $K_5Nd(MoO_4)_4$  and  $KNd_5(MoO_4)_8$ , *Journal of Luminescence* (2026) 121995. <https://doi.org/10.1016/j.jlumin.2026.121995>.
- M.U. Jamal**, V. Nagirnyi, K. Chernenko, A. Kotlov, Y. Smortsova, D. Spassky, Crystal structure controlled energy transfer to  $Tb_{3+}$  in  $KTb(MoO_4)_2$  and  $K_5Tb(MoO_4)_4$  crystals, *Materials Research Bulletin* 191 (2025) 113553. <https://doi.org/10.1016/j.materresbull.2025.113553>.
- D. Spassky, A. Vasil'ev, **M.U. Jamal**, V.A. Morozov, B.I. Lazoryak, B.S. Redkin, K. Chernenko, V. Nagirnyi, Temperature dependent energy transfer to  $Eu^{3+}$  emission centres in  $K_5Eu(MoO_4)_4$  crystals, *CrystEngComm* 26 (2024) 1106–1116. <https://doi.org/10.1039/D3CE01201H>.

**Konverentsiettekanded:**

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry, “Influence of crystal structure on energy transfer to  $Tb^{3+}$  in  $K_5Tb(MoO_4)_4$  and  $KTb(MoO_4)_2$  crystals”, Suuline ettekanne FMNT & NIBS konverentsil, oktoober 2024, Tartu, Eesti.

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry, “The influence of crystal structure on luminescence properties of  $Tb^{3+}$  in double molybdates”, posterettekannne 8. Balti elektrokeemia konverentsil, aprill 2024, Tartu, Eesti.

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry, “The impact of crystal structure on  $Tb^{3+}$  luminescence in double molybdates”, Posterettekannne LUMDETR 2024 konverentsil, juuni 2024, Riia, Läti..

**Jamal, Muhammad Usama;** Nagirnyi, Vitali; Spassky, Dmitry; “Luminescence properties of  $K_5Eu(MoO_4)_4$ ”, Posterettekannne ja 3-minutiline suuline ettekanne GSFMT teaduskonverentsil, 2023, Tartu, Eesti.

**Saadud uurimistoetused ja stipendiumid:**

2024 Saadud Eesti doktorkooli mobiilsustoetus osalemiseks LUMDETR 2024 konverentsil Riias, Lätis.

2020–2022 Magistriõpingute jooksul mitmel korral eriala- ja tulemusstipendiumide määramine.

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